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*Examples*

*for table*

Examples of pKa (acid/base) corresponding to the salt catalysts of the examples are given below:

ACID	ANION	pKa
HSCN	SCN <sup>⊖</sup>	0.85
HCOOH	HCOO <sup>⊖</sup>	3.8
C <sub>6</sub> H <sub>5</sub> COOH	C <sub>6</sub> H <sub>5</sub> COO <sup>⊖</sup>	4.2
CH <sub>3</sub> COOH	CH <sub>3</sub> COO <sup>⊖</sup>	4.8
C <sub>6</sub> H <sub>5</sub> SH	C <sub>6</sub> H <sub>5</sub> S <sup>⊖</sup>	7.8
CH <sub>3</sub> COCH <sub>2</sub> COCH <sub>3</sub>	CH <sub>3</sub> CO <sup>⊖</sup> CHCOCH <sub>3</sub>	9